

Unusual C-C Cleavage During Reduction of a β -Aminonitroalkene¹

Jeremiah P. Freeman^a, Lloyd Laurian^b and Jacob Szmuszkovicza^{a,b}

^aDepartment of Chemistry and Biochemistry, University of Notre Dame, Notre Dame, Indiana 46566

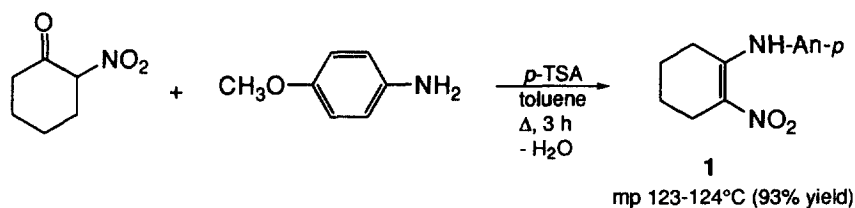
^bThe Upjohn Company, Kalamazoo, Michigan 49001

Received 22 February 1999; revised 15 April 1999; accepted 16 April 1999

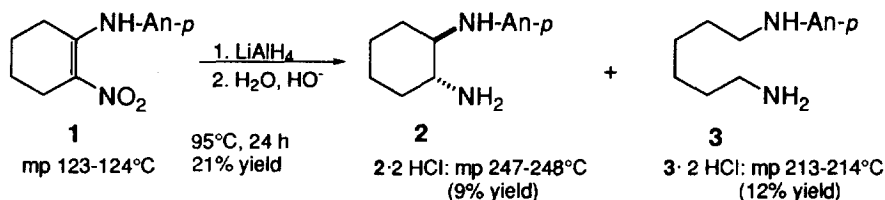
Abstract: An unusual ring cleavage product **3**, a diamine, was observed during LAH reduction of β -aminonitroalkene **1**. © 1999 Elsevier Science Ltd. All rights reserved.

Due to our long-standing interest in the diastereospecific formation of 1,2-diamines², we briefly examined the reduction of β -aminonitroalkenes (nitroenamines)³, previously suggested³ as a potential route to such compounds. While this method did not prove to be efficient in the cases we examined, due to side reactions, we did observe one unusual reduction mode.

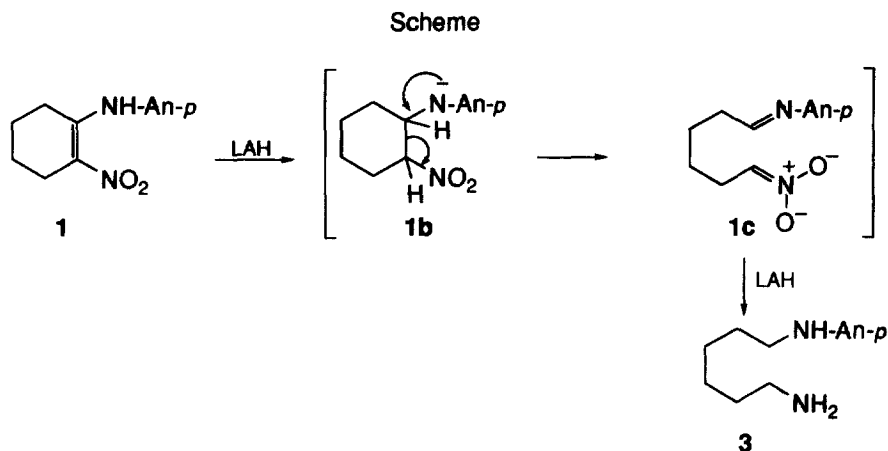
β -Aminonitroalkene **1** was prepared by condensation of *p*-anisidine with 2-nitrocyclohexanone⁴. Reduction of **1** with lithium aluminum hydride yielded two products: *trans*-diamine **2**



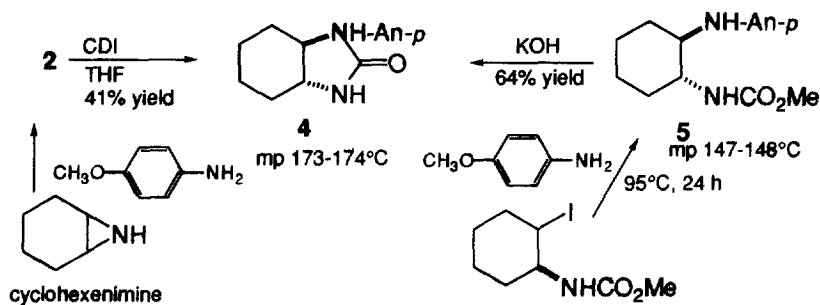
and the ring-cleavage product **3**. The structure of diamine **3** is based on its mass spectrum (M^+ 222) and its NMR spectrum.



We propose the reaction path shown in the Scheme for the transformation of **1** to **3**. Compound **1** undergoes conjugate reduction with LAH to the dihydro compound **1b**. A reverse aldol reaction brings about ring cleavage to **1c**, which is reduced by LAH to the observed product **3**.

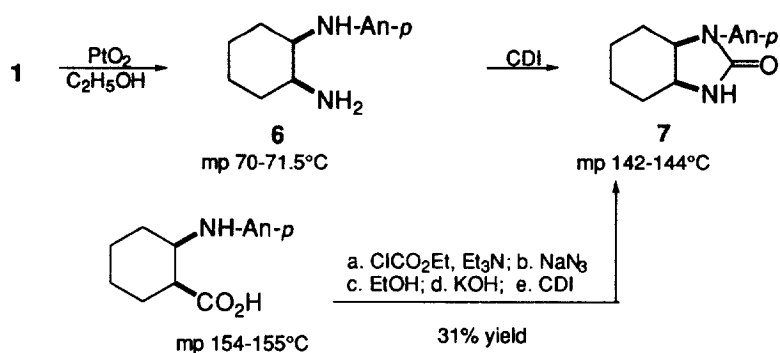


The stereochemistry of **2** was established by conversion to the corresponding imidazolidinone **4**, which was identical to that prepared from *trans*-1-(*N*-*p*-anisyl)-2-(*N*-carbomethoxy)cyclohexanediamine **5** by base-catalyzed ring closure. The diamine derivative **5** in

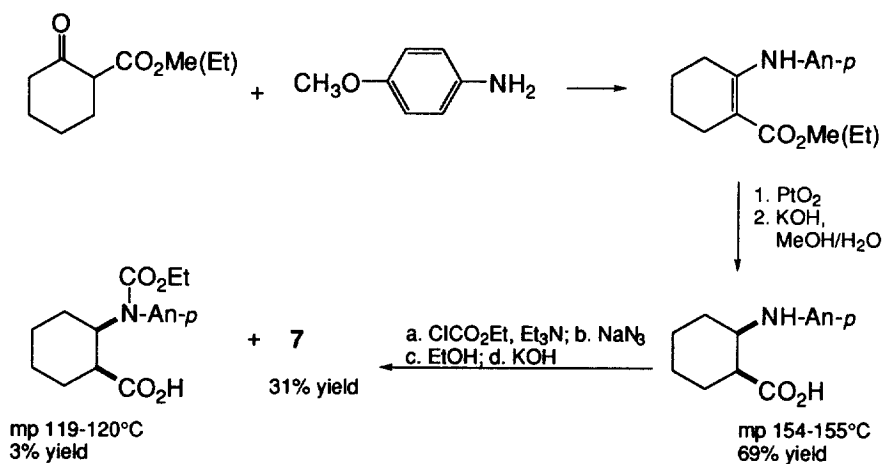


turn was prepared by the action of *p*-anisidine on methyl *N*-(*trans*-2-iodocyclohexyl)carbamate⁵. Diamine **2** could also be prepared in 21% yield by the action of *p*-anisidine on cyclohexenimine.

Catalytic reduction of **1** over platinum oxide produced *cis*-1-(*N*-*p*-anisyl)-2-cyclohexanediamine **6** in only an 8% yield along with *p*-anisidine in 48% yield. The stereochemistry of **6** was



consistent with its NMR spectrum⁶ and was established by conversion with carbonyl diimidazole to an imidazolidinone derivative 7,⁷ which was identical to that prepared independently from *cis*-2-(*p*-anisidino)-cyclohexanecarboxylic acid by the following steps:



Other nitroenamines analogous to 1 (*N*-phenyl and *N-p*-chlorophenyl) were subjected to catalytic reduction but only hydrogenolysis products could be identified. In particular no ring cleavage products were detected. LAH reductions were not examined.

References and Notes

1. The work described here was performed at the Upjohn Company (J.S. is now at the University of Notre Dame.)
2. Freeman, J.P.; Michalson, E.T.; D'Andrea, S.V.; Baczynskyj, L.; VonVoightlander, P.F.; Lahti, R.A.; Smith, M.W.; Lawson, C.F.; Scahill, T.A.; Miszak, S.A.; Szmuszkovicz, J. *J. Med. Chem.* **1991**, *34*, 1891. D'Andrea, S.V.; Freeman, J.P.; VonVoightlander, P.F.; Szmuszkovicz, J. *Tetrahedron*, **1991**, *47*, 6157.
3. Rajappa, S. *Tetrahedron*, **1981**, *37*, 1453.
4. Griswold, A.A.; Starcher, P.S. *J. Org. Chem.* **1966**, *31*, 357.
5. Swift, G.; Swern, D. *J. Org. Chem.* **1967**, *72*, 511.
6. Data for compound **6**: mp 70-71.5°C (ether). UV (EtOH): λ_{\max} 245.5 nm (11,850); 309.5 (2050). IR (nujol, cm^{-1}): 3360, 3260, 3180, 3100, 3030, 1620 w, 1575, 1535, 1510, 1260, 1250, 1240, 1180, 1035, 825. ^1H NMR (CDCl_3) δ 1.2-1.8 (8 H, m), 1.1-2.2 (3 H, m, exch.), 3.0-3.5 (2 H, m), 3.72 (3 H, s), 6.62 (2 H, AA', m), 6.77 (2 H, BB', m). Anal. Calcd for $\text{C}_{13}\text{H}_{20}\text{N}_2\text{O}$: C, 70.87; H, 9.15; N, 12.72. Found: C, 70.97; H, 9.24; N, 12.93. In a separate experiment a polymorph of compound **6** was isolated by virtue of its insolubility in ether. It melted at 106-110° and was identical by UV, IR and NMR to the compound described above.
7. Data for compound **7**: mp 142-144°C (ether). UV (EtOH): λ_{\max} 242 nm (11,800); sh 275 (1390); 280 (1500). IR (nujol, cm^{-1}): 3190, 3100, 3080, 1695, 1615, 1580, 1510, 1295, 1250, 1240, 1180, 1140, 1045, 855. ^1H NMR (CDCl_3) δ 1.12-1.9 (8 H, m), 3.72 (1 H, m), 4.08 (1 H, dt, $J = 6, 5, 5$ Hz), 3.8 (3 H, s), 5.54 (1 H, NH, s), 6.90 (2 H, m, AA'), 7.30 (2 H, m, BB'). Anal. Calcd for $\text{C}_{14}\text{H}_{18}\text{N}_2\text{O}_2$: C, 68.27; H, 7.37; N, 11.37. Found: C, 68.54; H, 7.39; N, 11.23.